

## *Directed A Grouping The Elements Answer Key*



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### Directed A Grouping The Elements

b. those in a group d. those in a horizontal row \_\_\_\_ 21. The physical and chemical properties of the elements change a. within a group. c. across each period. b. within a family. d. across each group. 22. For most elements, the has one or two letters, with the first letter always capitalized. 23. Horizontal rows of elements on the periodic table are called. 24.

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Grouping the elements. Metals, either one or two outer electrons, less reactive than alkaline earth metal, shiny good conductors of thermal energy and electric current; higher densities and melting points( except for Mercury) than elements in group 1 and 2. Do not give their electrons as easily as groups 1 & 2 making them less reactive.

### Grouping the elements Flashcards | Quizlet

Skills Worksheet Directed Reading A Section: Grouping the Elements Write the letter of the correct answer in the space provided. \_\_\_\_ 1. Which of the following gives elements in a family or group similar properties? a. the same atomic mass b. the same number of protons c. the same number of electrons in their outer energy level

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- Elements in a group often have similar properties because their atoms have the same number of electrons in their outer energy level.
- Hydrogen is set off by itself in the periodic table because its properties do not match the properties of any one group.
- Metals are in Groups 1-16. Metalloids are in Groups 13-16.

### Section 2 Grouping the Elements - Midway Middle School Science

Each element in the nitrogen group has electrons in the outer level. 22. Nitrogen from the air can react with what element to make ammonia for fertilizer? GROUP 16: OXYGEN GROUP. 23. How is oxygen different from the other four elements in Group 16? 24. The element can be found as a yellow solid in nature and is used to make sulfuric acid. 25.

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Directed Reading A Section: Grouping the Elements \_\_\_\_ 1. What gives elements in a family or group similar properties? a. the same atomic mass b. the same number of protons in their nuclei c. the same number of electrons in their outer energy level d. the same number of total electrons GROUP 1: ALKALI METALS \_\_\_\_ 2.

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Chapter 12, section 2 Grouping the Elements. This section is about the ways in which elements are grouped together. STUDY. PLAY. Alkali Metals (Group 1) Most reactive metals. Alkaline-Earth Metals (Group 2) Less reactive than the alkali metals. Transition Metals (Groups 3-12)

### Chapter 12, section 2 Grouping the Elements Flashcards ...

Section 2 Grouping the Elements Bellringer How do you know that a bird is a bird, that a kangaroo is a kangaroo, and that a shark is a shark? What characteristics of each animal help you to tell the animals apart? How can such an analysis of characteristics be applied to elements?

### Chapter 12 Section 2 Grouping the Elements Chapter 12

Directed Reading B Section: Grouping the Elements Circle the letter of the best answer for each question. 1. What do atoms of elements in a group have that makes their properties similar? a. the same atomic mass b. the same number of protons c. the same number of outer level electrons d. the same total number of electrons 2.

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Directed Reading A continued GROUP 13: BORON GROUP Use the terms from the following list to complete the sentences below. metal boron aluminum oxygen 1 2 . The most common element in Group 13 is , which is used to make airplane parts. 1 3 . Aluminum is useful because it is a lightweight . 1 4 .

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Directed Reading A Section: Grouping the Elements 1. What gives elements in a family or group similar properties? a. the same atomic mass b. the same number of protons in their nuclei c. e same number of electrons in their outer energy level d. the same number of total electrons GROUP 1: ALKALI METALS 2.

**janellmcclure.typepad.com**

Skills Worksheet Directed Reading B Section: Arranging the Elements 1. Why do you think scientists might have been frustrated by the organization of the elements in the early 1860s? DISCOVERING A PATTERN \_\_\_\_ 2. Which arrangement of elements did Mendeleev find produced a repeating pattern of properties? a. elements in order of increasing ...

**Skills Worksheet Directed Reading B - Welchclass.com**

Directed Reading A continued Atomic Number \_\_\_\_ 22. How many of the recently discovered elements follow the periodic law? a. half of them b. none of them c. every fifth element d. all of them \_\_\_\_ 23. What does the periodic law say about the properties of elements? a. They change periodically with the atomic numbers of the elements. b.

**Skills Worksheet Directed Reading A - RCSMSMrs. Botten**

Each element in the Nitrogen Group has electrons in the outer level. 23. Nitrogen from the air can react with what element to make ammonia for fertilizer? GROUP 16: OXYGEN GROUP 24. How is oxygen different from the other four elements in Group 16? 25. The element can be found as a yellow solid in nature and is used to make sulfuric acid. 26.

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