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Ncert Chemistry 12 Intext Answer

3.7 Define conductivity and molar conductivity for the solution of an electrolyte. Discuss their variation with concentration. Sol. Conductivity of a solution is defined as the conductance of a solution of 1 cm length and having 1 sq. cm as the area of cross-section. Molar conductivity of a solution at a dilution (V) is the conductance of all the ions produced from one mole of the electrolyte ...

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NCERT Solutions Class 12 Chemistry UNIT 1 THE SOLID STATE Intext Questions. NCERT Solutions, CBSE Sample paper, Latest Syllabus, NCERT Books, ... Question 1.12: Distinguish between (i) Hexagonal and monoclinic unit cells (ii) Face-centred and end-centred unit cells.

NCERT Solutions Class 12 Chemistry UNIT 1 THE SOLID STATE ...

NCERT Solutions For Class 12 Chemistry Chapter 10 Haloalkanes and Haloarenes. NCERT Solutions For Class 12 Chemistry Chapter 10 Haloalkanes and Haloarenes. RD Sharma. ... Explain your answer. Ans. In S_N2 mechanism, reactivity depends upon the steric hindrance around the C-atom carrying the halogen. Lesser the steric hindrance, faster the ...

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Class 12 Chemistry. Solution In Text Questions 1. Question - 2.1 - Calculate the mass percentage of benzene (C_6H_6) and carbon tetrachloride (CCl_4) if 22 g of benzene is dissolved in 122 g of carbon tetrachloride.. Answer: Given, Mass of solute (W B) = 22g Mass of solvent (W B) = 122 g. Mass of solution = 22g + 122 g = 144 g

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